



Massachusetts Institute of Technology  
Center for Environmental Health Sciences

Sponsored by NIH-NIEHS Core Center P30-ES002109 and the Department of Biological Engineering

## 2017 Center for Environmental Health Sciences Poster Session

Wednesday, Jan. 18, 2017  
2:30pm – 5:00pm  
Lobby of Building 13

Abstract Submission Deadline: [December 2, 2016](#)

The Abstract Form can be found via the CEHS

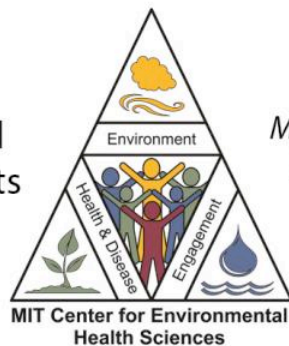
<https://cehs.mit.edu/events/poster-session-2017>

Prizes will be given for the Best Posters  
in Engineering and Life Sciences.

1<sup>st</sup> Place: \$1,000

2<sup>nd</sup> Place: \$500

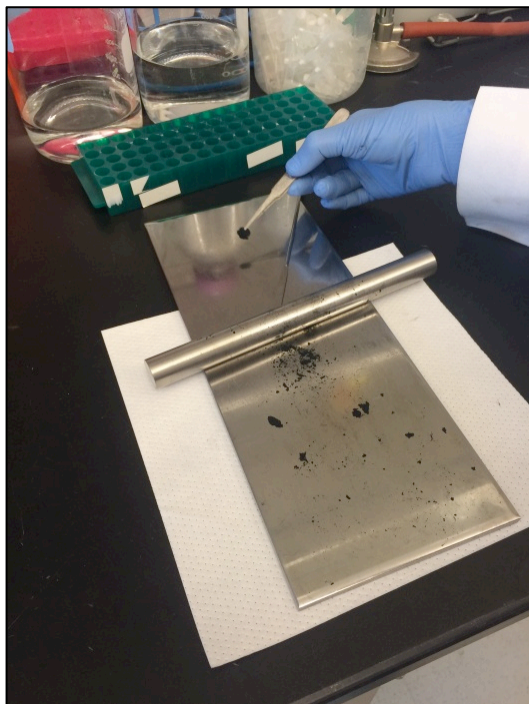
3<sup>rd</sup> Place: \$200 and  
CEHS gifts



*Supported by the  
Myriam Marcelle Znaty  
Memorial Research*

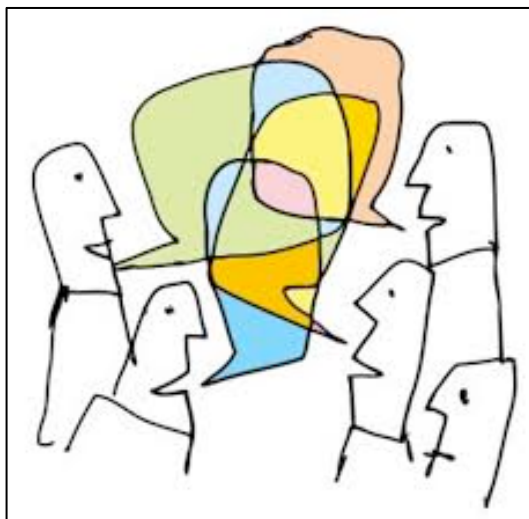
# M3D3: Cathode construction

11/30/2016



# Today in lab

- ✓ Quiz
- Prelab discussion
- In 76-591: prepare material and roll & punch the cathodes
- In 56-322: research proposal peer review exercise
- ❖ Reflect on elevator pitch to Prof. Belcher



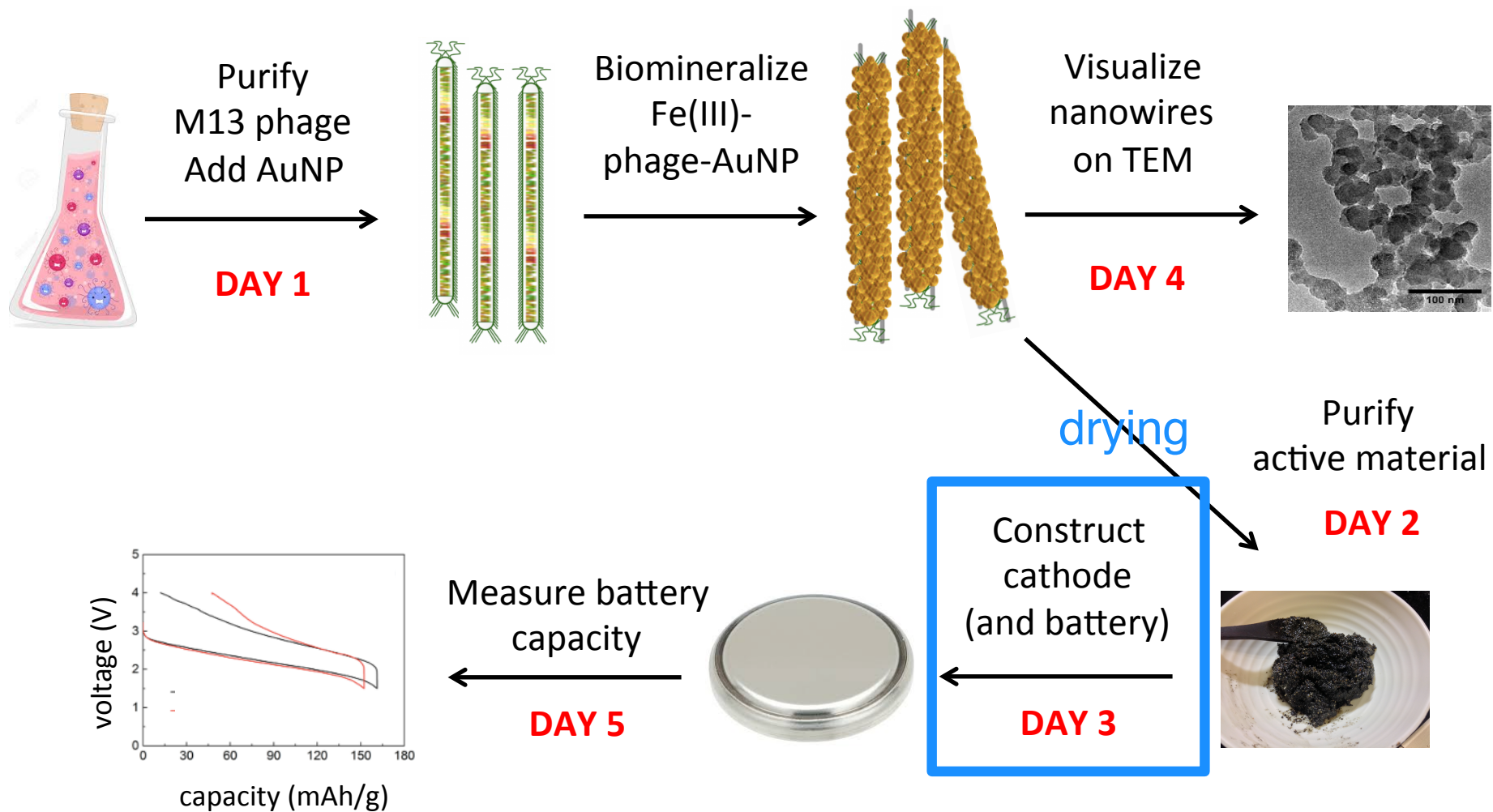
# On the horizon for M3

- Major assignments **do not change ideas after 12/01 at 5pm**
  - **Research proposal oral presentation** (20%); Friday, December 9, 1pm
  - **Mini-report** (5%); Monday, December 12, 10pm
- Homework due M3D4
  - Submit both parts as a **team**
  - 1. Refine **presentation outline**, incorporate peer review feedback
  - 2. **Background and Approach**, with references  
<http://belcherlab.mit.edu/publications/>

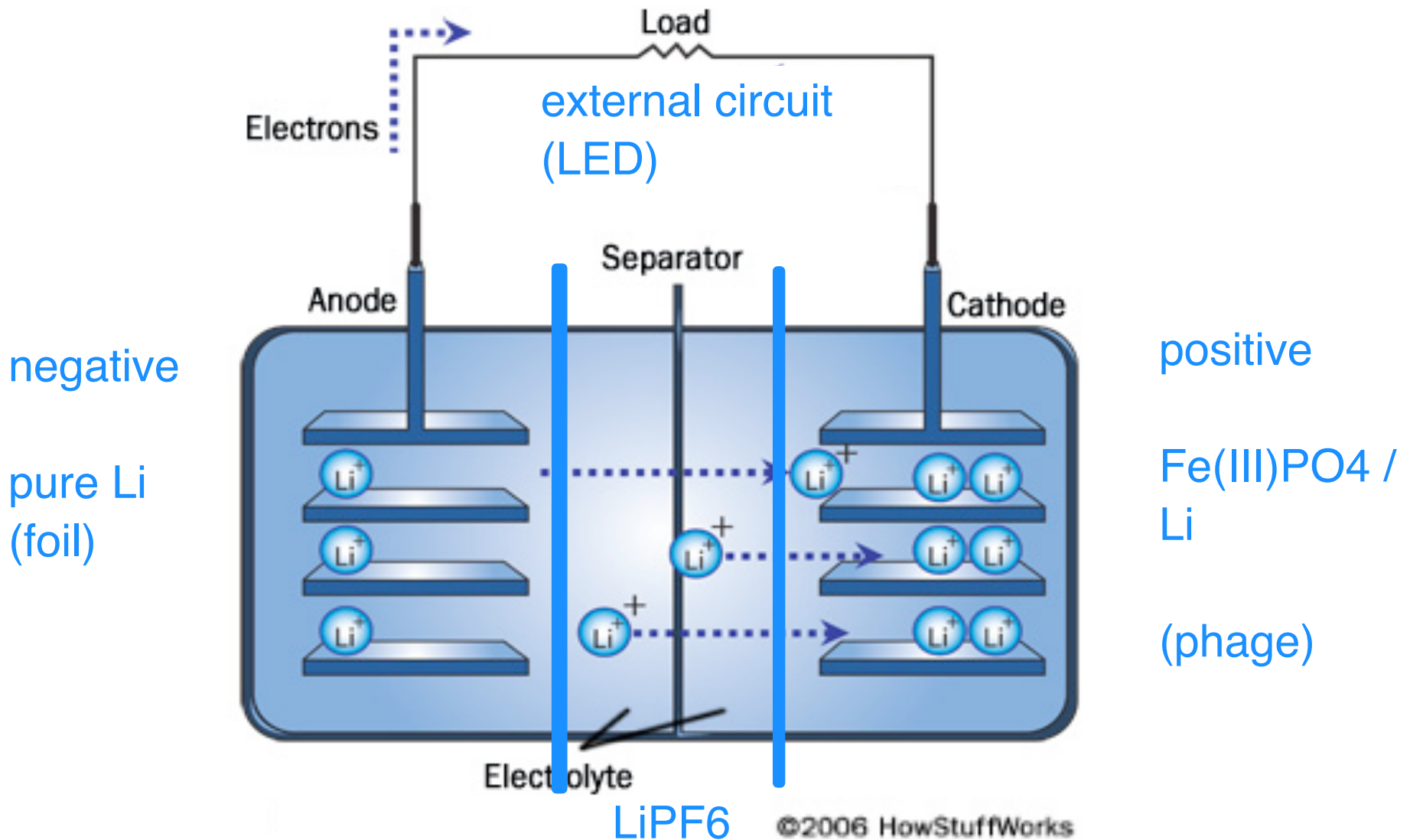
- Visit



# Module 3: biomaterials engineering overview

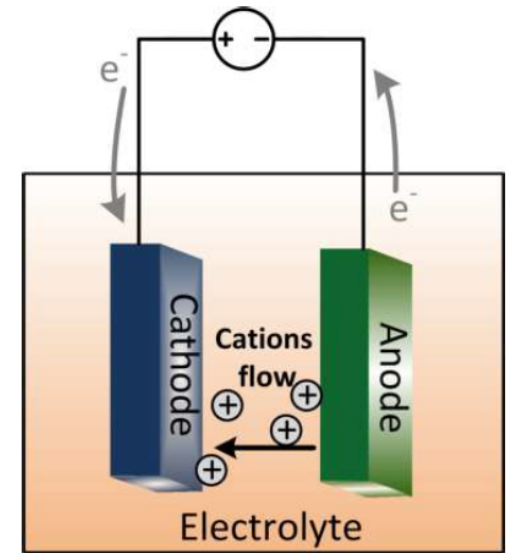


Is this battery **discharging** or charging?



# Main components of a battery

- During **discharge**,
  - cathode accepts electrons  $e^-$  and lithium ions  $Li^+$
- Battery consists of two electrodes:
  - cathode = positive electrode, **accepts** electrons
  - anode = negative electrode, **gives** electrons
  - In rechargeable battery, when is electrode polarity defined? **discharge**
  - **electrolyte** allows for flow of ions
- What is **capacity**?
  - quantity of electricity (charge) involved for the electro-chemical reaction within the battery
  - for our Fe(III)-phage batteries, the theoretical capacity is 178 mA\*h/g



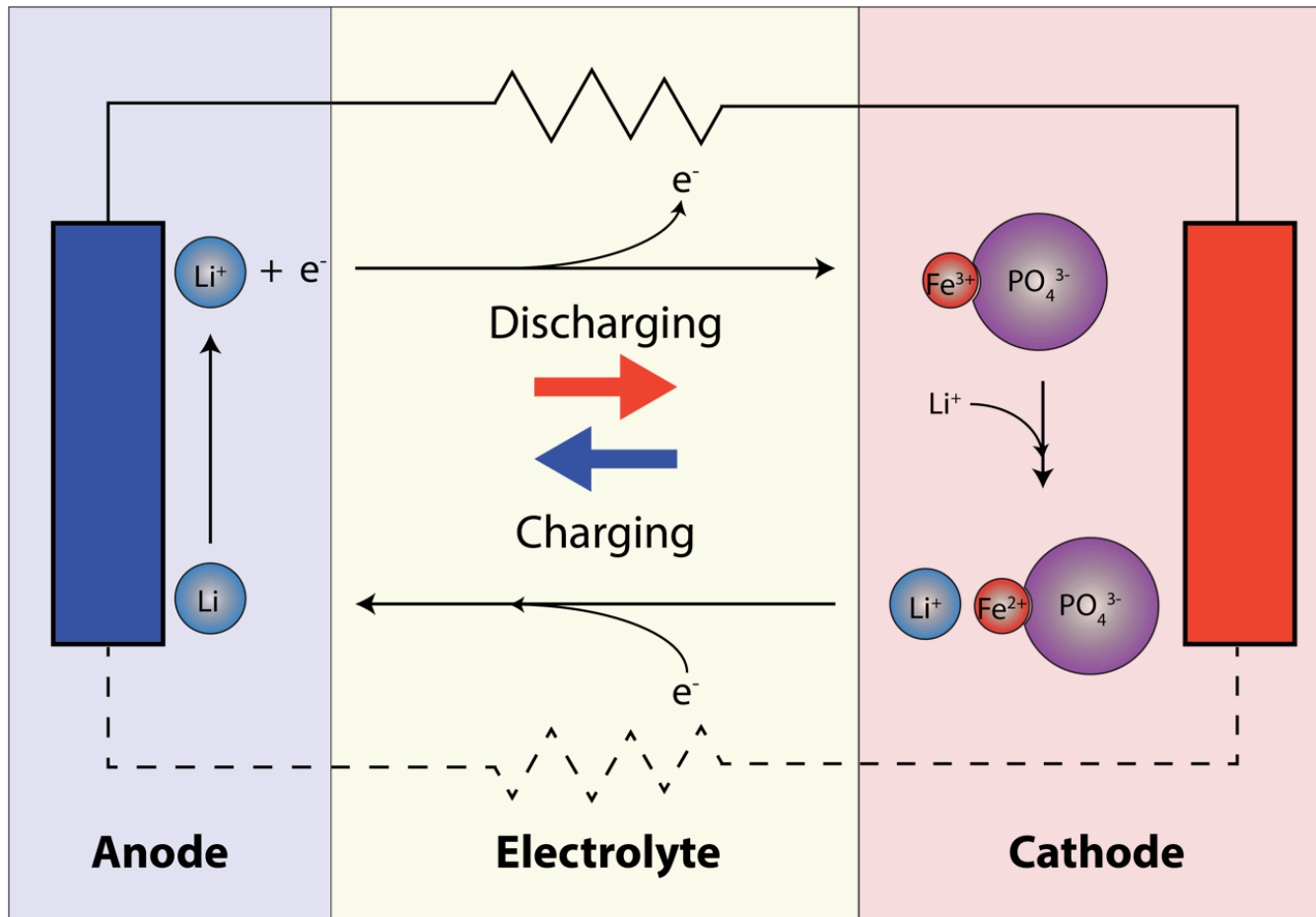
from Dr. Maryam Moradi

# Diagram of M3 battery

M13 phage

AuNP

Fe(III) PO<sub>4</sub> / Li Fe(II)PO<sub>4</sub>

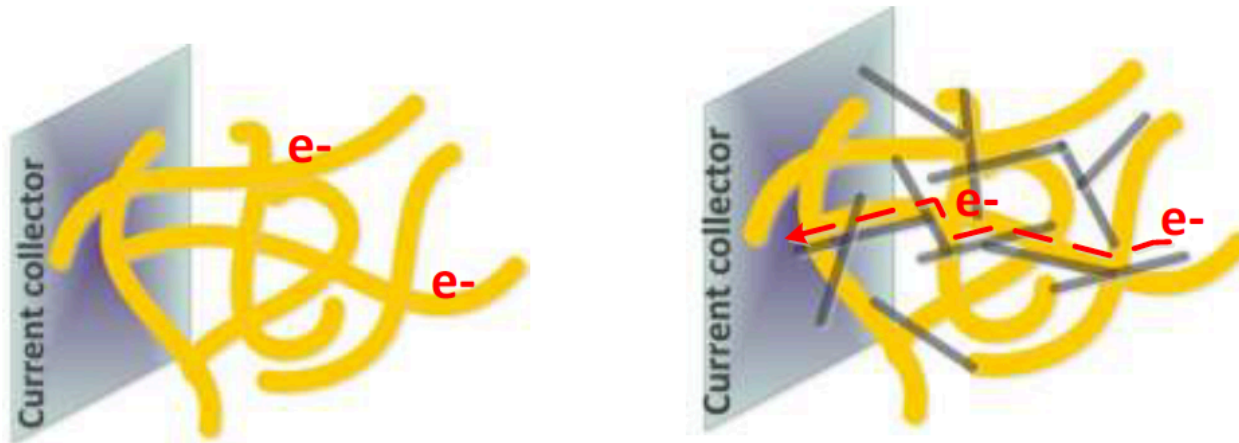




# How can a phage scaffold improve battery?

- ion diffusivity → nano structuring active material surface/volume higher
- electronic conductivity → integrating additives (“doping”) screening by phage display

Example: adding carbon nanotubes to phage cathode



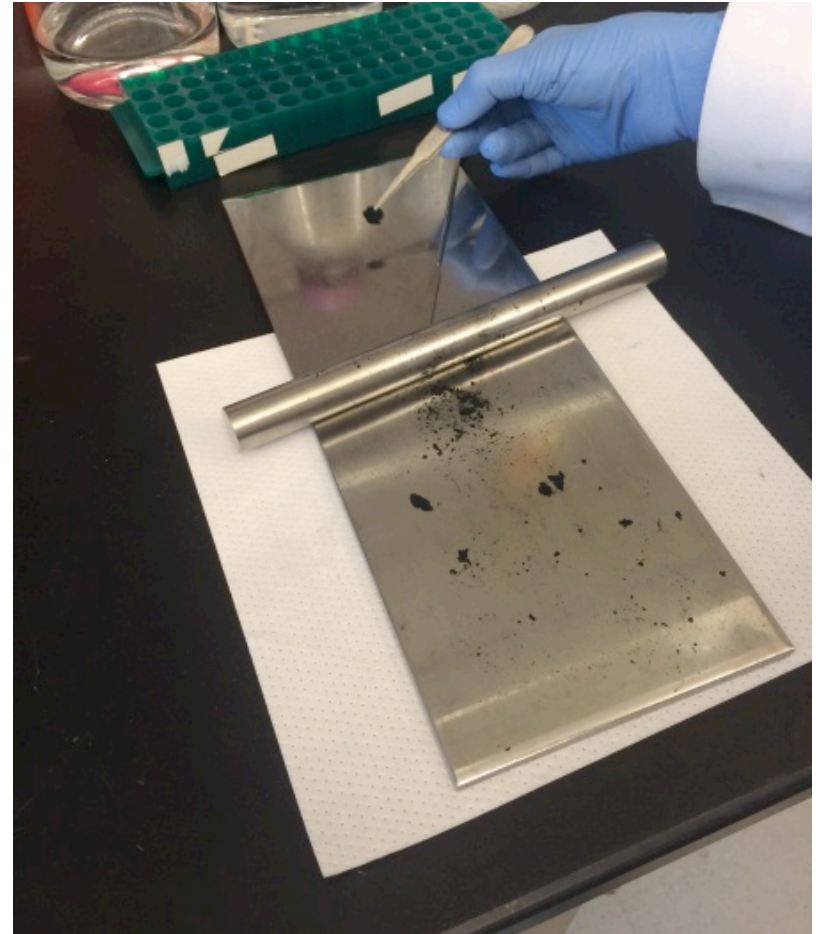
from Dr. Moradi, Belcher Lab

- How might AuNP size affect your battery capacity?
  - experimental variables:

4 nm or 9 nm

# How will you construct your cathode?

1. Weigh Fe(III)-phage-AuNP nanowires (active material)
2. Mix with Super P: **carbon** and PTFE: **teflon binder**
3. Roll material into thin sheet
4. "Punch out" cathode disc
5. Weigh cathode
6. Dry cathode



## One large W/F team:

- Part 1: cathode construction in the Belcher Lab
- Part 2: peer review in the 20.109 Lab